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Synthesis and Cytotoxic Activity of New 4'-Deoxy C-3'-Homo Anthracyclines

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Abstract: Starting from (S)-propylene oxide, the protected C-3 homolog of 4-deoxy daunosamine 11 is prepared in 4 steps. Acid catalyzed glycosidation of anthracyclinones such as daunomycinone, β -rhodomycinone and ε isorhodomycinone with 11 affords new analogs of anthracyclines demonstrating strong cytotoxicity against L1210 leukemia, A-549 and HT-29 tumor cells.

INTRODUCTION

The clinical efficiency of anthracycline antibiotics, and particularly of doxorubicin $1a₁$ in anticancer chemotherapy has induced numerous studies to improve the therapeutic index of this class of compounds. In this respect, modification of the sugar moiety of anthracyclines has been particularly fruitful leading to improved analogs such as epirubicin (4'-epi doxorubicin) $2,2$ pirarubicin (4'-0-tetrahydropyranyl doxorubicin) $3³$ or esorubicin (4'-deoxy doxorubicin) 4.⁴

1a Doxorubicin $R=R_1=OH$, $R_2=H$ 1b Daunorubicin $R = R_2 = H$, $R_1 = OH$ 2.4'-epi doxorubicin $R=R_2=OH$, $R_1=H$ 3 Pirarubicin R=OH, R_1 =OTHP, R_2 =H 4 Esorubicin R=OH, $R_1=R_2=H$

6a R=R₁=H, R₂=OH 6b $R = R_2 = OH$, $R_1 = H$

All these compounds display a reduced cardiotoxicity compared to **1** in clinical trials. Other modifications giving highly active analogs, not implying the C-4' hydroxyl, have been described: replacement of the C-3' NH₂ group by OH,⁵ introduction of a 2'-iodo⁶ or 2'-fluoro⁷ substituent or alkylation of the primary amine function⁸ (including the formation of the potent morpholino analogs⁹).

A new type of modification has been suggested by $Pulman¹⁰$ on the basis of intercalation models of daunorubicin 1b with the double-stranded oligonucleotide d(CGATCG)₂. Since there is a correlation between the stability of the DNA-anthracycline complex and antitumor activity of the corresponding anthracycline, these authors have suggested the synthesis and evaluation of new analogs, such as **5a,b, bearing** a -CH2NH2 group instead of the amino group at $C-3'$. The increase in affinity estimated as δ (difference of the resulting energy balance ΔE , which is the sum of the intermolecular interaction energy (ΔE_{inter}) and of the conformational energy change of the anthracycline (ΔE_{conf}), between this analog and **lb**) is estimated to be as high as 15 kcal/mol and may be due to better electrostatic interactions of the protonated amine (more imbedded in the minor groove) with the DNA backbone.

Subsequently, a patent¹¹ has described the synthesis of analogs such as $6a$,b (including the C-3' epimers) in the 4'-epi series using as the key step the condensation of nitromethane with methyl-2,6dideoxy- α -L-erythro-hexapyranosid-3-ulose followed by reduction of the corresponding nitroalkene. However only the cytotoxicity and antitumor activity of the $C-3'$ epimer (prepared by silver triflate catalyzed condensation of the corresponding 1-chloro derivative) has been reported.

Taking into account this hypothesis and the observation that esorubicin 4 is more active and less cardiotoxic than 1a, the synthesis of 4'-deoxy C-3'-homoanthracyclines seems worthwhile and thus the preparation of the suitable carbohydrate derivative 12 was first carried out followed by the glycosidation of known anthracyclinones, such as daunomycinone, β -rhodomycinone and ϵ -rhodomycinone.

PREPARATION OF METFIYL GLYCOSIDE 12.

Commercially available cyanopropionaldehyde diethyl acetal 7 is treated with LDA in THF at -78°C and the resulting anion is then reacted with (S) -propylene oxide (1.3 eq) to give an unseparable mixture of the two epimeric hydroxy nitriles 8a,b in 98% isolated yield. HPLC analysis (Lichrosorb Si 60 7µ, eluent hexane-AcOEt, $7:3$) shows that the a/b ratio is $40/60$ and this is in agreement with the observation of signals at δ 2.89 and 3.05 ppm (in similar relative intensity), corresponding to the α -cyano proton.

Methanolysis of this mixture affords four compounds which are separated by chromatography over silica to give by decreasing polarity order: 10a (41%) , $9b$ (7.5%) , 10b (19%) and 9a (32%) . The overall ratio of 10a,b/9a,b is similar to the 60/40 ratio observed for 8a,b thus allowing the structure determination of the major nitrile formed in the epoxide condensation. Structures of the four methyl glycosides are consistent with their 1 H NMR spectra (Table 1).

*: partly hindered, sext.: sextuplet

As expected the amount of the more stable α anomers (9a and 10a) accounts for 73% of the mixture. The configuration of the nitrile group is easily determined by the signal of the corresponding proton and by the relative deshielding of the C-5 proton in the case of the axial isomers (CN anisotropy).

However the amount of the desired isomer 9a may be increased by base-catalyzed equilibration of the methanolysis mixture prior to chromatography, a stable 80/20 ratio of 9a,b/10a,b being reached after treatment with tBuOK (0.4 eq.) in tBuOH for 12 h at room temperature.

LiAlH₄ reduction of nitrile 9a affords the primary amine 11 (91%) as a yellow oil. This material could not be converted to trifluoroacetamide 12 under the standard trifluoroacetic anhydride-pyridine conditions since no material could be recovered probably due to extensive decomposition. However treatment of 11 with methyl trifluoroacetate in presence of DBU affords the desired amide 12 in 76% yield.

GLYCOSIDATION

Glycosidation reactions of anthracyclinones have been reported under different conditions, including the Koenigs-Rnorr procedure using 1-bromo 2-deoxy carbohydrates, readily prepared with HBr from the corresponding 1-O-acyl derivative or glycal, or by direct Lewis or protonic acid-catalyzed condensation of the 1-O acyl derivative. However all attempts to prepare these derivatives failed in our case. Acid catalyzed hydrolysis of 12 affords a 80/20 mixture of the α and β anomers 13a and 13b in 95% yield from which the corresponding acetates could be obtained (Ac₂O, cat. H₂SO₄, 20 $^{\circ}$ C, 48h) in only 30% isolated yield.¹² Furthermore attempted treatment of the latter with HBr to get the 1-bromo derivative or thermal or SiO₂catalyzed elimination of acetic acid to afford the corresponding glycal failed.

These disappointing results, and particularly the easy formation of by-products, probably arise from the high reactivity of the hemiacetal together with the possible participation of the -CH₂NHCOCF₃ group. The observation that the tetrahydropyran ring bears no oxygenated group at $C-2$ or $C-4$ as in carbohydrates and that the 3-NHCOR is replaced by a less deactivated group -CH2NHCOR clearly suggests that the formation of the corresponding oxonium ion is easier in this case.

These considerations prompted us to attempt an acid-catalyzed transglycosidation of the methyl glycoside 12 with anthracyclinones in presence of 4 Å molecular sieves as methanol trap. Reactions were carried out in anhydrous CH₂Cl₂ in presence of anhydrous PTSA (increasing amounts 0.1 to 0.5 eq.) using 1 eq. of 12 and Daunomycinone 14, β -Rhodomycinone 15 and ε -isorhodomycinone 16. In each case a single glycoside 17-19 is obtained in isolated yields ranging from 10-1696 (-60% based on recovered aglycones). The structures of these new glycosides are consistent with their MS and ${}^{1}H$ NMR spectra. For example, the signals for H-1 and H-7 appear respectively at δ 5.54 and 5.31 ppm in the case of 17 in good agreement with the chemical shifts observed for 4-deoxydaunomycin (δ 5.59 and 5.30 ppm). Similar values are observed in the case of 19 (δ 5.49 and 5.25 ppm) and of 18 (δ 5.47 and 5.14 ppm). In the case of the β -Rhodomycinone, the formation of a 10-O-glycoside instead of the less hindered 7-O-glycoside 18 may be ruled out on the basis of the chemical shift for H-10 in 18 (δ 4.85 ppm) and β -rhodomycinone (δ 4.87 ppm) which is lower than for 10-O-glycosides (i.e. δ 4.99 ppm in the case of a 7,10-bis- α -L-fucopyranosyl derivative¹³).

CYTOTOXIC ACTIVITY

Compounds 17-19 were tested *in vitro* against L1210 leukemia, A-549 and HT-29 tumor cells (MTT reduction Assay) as well as compounds 20 and 21 obtained by base-catalyzed hydrolysis of 17 and 18 (immediately before testing). The results summarized in Table 2 show that (the ε -isorhodomycinone glycoside 19 is much less active with IC₅₀'s higher than 1 μ g/mL) the glycosides obtained from β -Rhodomycinone and Daunomycinone are highly active and that glycoside 20 demonstrate an activity in the same range as Doxorubicin (the C-3' epimer of 6a has also been reported¹¹ to display IC₅₀ values against HeLa and P388 cells similar to those observed for daunorubicin).

Compound	L ₁₂₁₀	$A - 549$	$HT-29$
Doxorubicin	0.02	0.05	0.06
17	0.25	0.67	0.43
18	0.14	0.48	0.14
20	0.04	0.10	0.04
21	0.11	0.13	0.13

Table 2. Cytotoxicity (IC ς_0 in μ g/mL) in the MTT assay.¹⁴

CONCLUSION

Starting from the hypothesis of Pullman, new 4'-deoxy anthracyclines bearing a 3'-CH₂NHR group instead of the usual aminogroup have been prepared from (S)-propylene oxide. The preliminary results obtained for the in vitro activity against L1210, A-549 and HT-29 seems to substantiate the initial hypothesis. Further experiments such as DNA binding measurements and in vivo activity tests are however needed to precise the biological properties of this new kind of glycosides.

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EXPERIMENTAL

Melting points were measured with a Tottoli Buchi 510 apparatus and are uncorrected. lR spectra were recorded on a Beckman, Acculab 2 spectrophotometer. ¹H NMR spectra were obtained on a Brucker WP 200 SY spectrometer using CDCl₃ as a solvent and TMS as as internal standard. Microanalyses were obtained from the "Service Central de Micmanalyses" (CNRS, Lyon).

3(R,S)-cyano 1,1-diethoxy 5(S)-hexanol 7a,b.

To a solution of diisopropylamine (4.6 mL, 32.8 mmol) in anhydrous THP (32.4 mL) (kept at -10°C under $N₂$) was slowly added nBuLi (20.5 mL of a 1.6 M solution in hexane). After stirring the resulting colourless mixture at -10° C during 30 min and then cooling at -78° C, a solution of 3-cyanopropionaldehyde diethylacetal (4.32 g, 275 mmol) in **anhydrous THF (2** mL) was slowly added and stirring was then continued for 2 h at -78°C. At the same temperature, propylene oxide (2.04 g, 1.3 eq.) was added and the mixture was stirred overnight while slowly warming to room temperature. The reaction mixture was then pourred into a solution of saturated $NH₄Cl$ and extracted with ether. The combined extracts were dried over Na₂SO₄ and after removal of the solvent, the residue was rapidly filtered over $SiO₂$ (eluent : hexane-AcOEt, 7:3) to give **7a, b(5.8 g, 98%) as** a yellow oil which was used in the next step without further purification. ¹H NMR (CDCl₃): δ 1.23 (9H, t), 1.68 (2H, m), 1.89 (2H, m), 2.28 (1H, m), 2.89 (0.5H, p, J=6.5Hz), 3.05 (0.5H, h, J=4.5 Hz), 3.65 (4H, m), 4.01 (lH, m), 4.70 (lH, t) ppm.

Methyl glycosides 8a,b and 9a,b.

To a solution of **7a,b** (2.55 g, 11.8mmol) in anhydrous MeGH (12.5 mL) was added, at room temperature and under a N_2 atmosphere, a 0.92N solution of HCl in anhydrous MeOH (13 mL). The reaction was stopped when TLC (eluent : hexane-AcOEt, 4:l) indicated consumption of starting material (16 h) and the formation of four new compounds. After hydrolysis with a saturated NaHCO3 solution and extraction with ether, the combined ether phases were washed with water and dried over $Na₂SO₄$. Evaporation under reduced pressure of ether, afforded a residue whose chromotography over $SiO₂$ (progressive elution with hexane-AcGEt, 98:2 to 80:20) yielded successively 9a (32%), **lob** (19%). **9b** (7.5%) as oils and **lOa** (4196, mp 61°C). The overall yield of the cyclisation was 82% from **8a,b.**

Note: Ethyl glycosides are detected by NMR at shorter reaction time.

Methyl 3-cyano-2, 3, 4, 6-tetradeoxy-α-L-lyxohexopyranoside 9a.

oil; $[\alpha]_D^{20}$ -126 (c 0.65, CHCl₃); ¹H NMR (CDCl₃) : 1.2 (3H, d, J=7Hz, H-6), 1.53 (1H, q, J=11.5 Hz, H-4ax), 1.82 (1H, td, $J_1=11.5$ Hz, $J_2=3.5$ Hz, H-2ax), 2.0 (2H, m, H-2eq and H-4eq), 3.08 (1H, tt, J_1 =3.5Hz, J_2 =11.5Hz, H-3), 3.33 (3H, s, OMe), 3.82 (1H, m, H-5), 4.76 (1H, s, W_H=6Hz, H-1) ppm; I.R. (CHCl3) : 2220 cm⁻¹ (CN). Anal.: calc. for C₈H₁₃NO₂: C 61.91, H 8.41; found: C 61.64, H 8.49. NMR and IR data of the other isomers are given below:

Methyl 3-cyano-2, 3, 4, 6-tetradeoxy-β-L-*ribo*hexopyranoside 10b.

¹H NMR (CDCl₃): 1.3 (3H, d, J=6Hz, H-6), 1.36 (2H, m, H-4, H-2), 1.84 (1H, dd, J₁=9Hz, J₂=2Hz, H-2), 2.04 (1H, dd, J₁=9Hz, J₂=2Hz, H-4), 3.17 (1H, s, W_H=10Hz, H-3), 3.52 (3H, s, OMe), 3.92 (1H, sext.d, J_1 =7.5Hz, J_2 =2Hz, H-5), 4.75 (1H, dd, J_1 =9.5Hz, J_2 =2.5Hz, H-1) ppm; 1.R. (CHCl₃) : 2240 cm^{-1} (CN).

Methyl 3-cyano-2, 3, 4, 6-tetradeoxy-β-L-lyxo hexopyranoside 9b

¹H NMR (CDCl₃): 1.28 (3H, d, J=6Hz, H-6), 1.44-1.61 (2H, m, H-4, H-2), 1.95 (1H, md, J=12.5Hz, H-4), 2.05 (1H, md, J = 12.5Hz, H-2), 2.75 (1H, tt, J₁ = 5Hz, J₂ = 12Hz, H-3), 3.50 (3H, s, OMe), 4.27 (dd, J_1 =9.5Hz, J₂=2.5Hz, H-1) ppm; I.R. (CHCl₃): 2240 cm⁻¹ (CN).

Methyl 3-cyano-2, 3, 4, 6-tetradeoxy-α-L-ribohexopyranoside 10a

mp 61°C; $\alpha \ln^{20}$ -152 (c 1.19, CHCl₃); ¹H NMR (CDCl₃): δ 1.21 (3H, d, J=6Hz, H-6), 1.56 (1H, m,), 1.81 (2H, m,), 2.0 (1H, m,), 3.03, (1H, s, W_H=11.5Hz, H-3), 3.38 (3H, s, OMe), 4.18 (1H, sext.d, J_1 =6Hz, J_2 =2Hz, H-5), 4.76 (1H, s, W_H=6Hz, H-1) ppm; I.R. (CHCl₃): 2260 cm⁻¹ (CN).

Base catalyzed equilibration of 10a

A solution of 1Oa (O.l26g, 0.82 mmol) in dry tBuOH (2 mL) was stirred in presence of tBuOK (0.036g 0.4 eq.) at room temperature for 12h. HPLC analysis (column: Lichrosorb Si-60, 7μ , eluent: hexane-ethyl acetate, 70/30, V/V) gives a **9a/lOa ratio** of 82/18. Identical figures were obtained after 72h at room temperature or at reflux.

Methyl 3-aminomethyl-2, 3, 4, 6-tetradeoxy-a-L-lyxohexopyranoside 11

To a suspension of LiAlH₄ (0.23 g) in anhydrous THF (8.4 mL) was added slowly a solution of 9a (0.83 g, 5.33 mmol) in THF (16.4 mL). After stirring the reaction mixture for 4 h at room temperature, TLC indicated consumption of starting material. Hydrolysis was performed by slow addition of a 10% solution of KOH (1.3 mL). The precipitated inorganic salts were filtered off and washed with AcOEt. Evaporation of the solvent afforded amine **11 (0.77 g, 91%) as** a yellow oil which was used in the next step without purification.

¹H NMR (CDCl3): 0.90 (1H, q, J = 12Hz, H-4ax), 1.17 (3H, d, J = 7Hz, H-6), 1.31 (2H, m), 1.8 (3H, m), 2.51-2.55 (2H, dd, J_1 = 7Hz, J_2 = 2Hz, CH₂N), 3.35 (3H, s, OMe), 3.85 (1H, m, H-5), 4.77 (1H, s, H-l) ppm.

Methyl N-trifluoroacetyl 3-aminomethyl-2, 3, 4, 6-tetradeoxy-a-L-lyxo hexopyranoside 12

The amine **11 (0.77 g, 4.&1** mmol) was dissolved in methyl trifluoroacetate (27 ml, 55 eq.) in presence of DBU (0.26 mL, 0.35 eq.). The reaction mixture was stirred at room temperature overnight under N₂. After removal of CF₃COOMe under reduced pressure, chromatography over SiO₂ (Eluent: 3% MeOH-CH₂Cl₂) of the crude product afforded 12 (0.94 g, 76%) as a white powder.

mp 67°C; $[\alpha]_D^{20}$ -90.4 (c 0.5, CHCl₃); ¹H NMR (CDCl₃): δ 1.01 (1H, q, J = 11.5Hz, H-4ax),1.19 (1H, d, $J=$ 7Hz, H-6), 1.29 (1H, td, $J_1=$ 3Hz, $J_2=$ 11.5Hz, H-2ax), 1.70 (2H, m, H-4eq and H-2eq), 2.20 (1H, broad s, W_H = 23Hz, H-3), 3.20 (2H, t, J = 6Hz, CH₂NH-), 3.34 (3H, s, OMe), 3.85 (1H, sept., J = 6Hz, H-5), 4.77 (1H, broad s, W_H = 7Hz, H-1), 7.21 (1H, broad s, W_H = 14Hz, NH) ppm; I.R. (CHCl₃): 3440 (NH), 3020, 2980 and 2940 (C-H), 1720 (C=O), 1540 (NH), 1450, 1385, 1225, 1160 and 1120 (C-O) cm⁻ ¹; Anal.: calcd. for $C_{10}H_{10}NO_3F_3$: C 47.06, H 6.32; found: C 47.33, H 6.21.

GeneraI **procedure for the glycosidation reaction.**

To a solution of the aglycone (0.26 mmol) in anhydrous CH₂Cl₂ (40 mL) kept under nitrogen, was added 4 **A** molecular sieves (0.1 g) and anhydrous FTSA (0.1 eq.). The resulting mixture was then stirred for 30 min. A solution of methyl glycoside (1 eq.) in anhydrous CH_2Cl_2 (2 mL) was then added dropwise and the reaction was followed by TLC (eluent 3% MeOH-CH₂Cl₂). After 3 h, another 0.5 eq. of PTSA was added while stirring was continued for an additionnal 24 h. The mixture was hydrolyzed and extracted with $CH₂Cl₂$. The combined organic layers were washed successively with a saturated NaHCO₃ solution and water. After usual workup, the aglycone-glycoside mixture was separated on preparative plates (eluent 2%) $MeOH-CH₂Cl₂$).

(7S,9R)-7-[[3-(Trifluoroacetamidomethyl)-2,3,4,6-tetradeoxy-a-L-lyxo-hexopyranosyl]oxy]-7,8,9,10tetrahydro-6,9,11-trihydroxy-9-acetyl-4-methoxy-5,12-naphtacenedione 17

mp 84-85°C; $[\alpha]_D^{20}$ 206.9 (c 0.16, CHCl₃); ¹H NMR (CDCl₃): δ 1.25 (m, 3H), 2.42 (s, 3H), 2.9 (d, J= 15 Hz, 1H), 3.23 (t, J = 7Hz, 2H), 4.06 (s, 3H), 4.78 (s, 1H), 5.31 (s, 1H), 5.54 (s, 1H), 6.36 (broad s, 1 H), 7.41 (d, J= 8H2, lH), 7.78 (t, J= 8H2, lH), 8 (d, J= 8Hz, Hi), 13.28 (s, lH), 13.98 (s, HI) ppm; I.R. (CH_2Cl_2) : 3680 and 3420 (OH), 3060 (CH), 1730 (C=O), 1620 and 1580 (C=C), 1260, 1210 and 1170 (C-O) cm⁻¹. MS (70 eV): 621, 399, 398, 382, 380, 363, 362, 339, 337, 321, 309, 301. HRMS calcd. for C₃₀H₃₀O₁₀NF₃ 621.18215 found 621.18218.

(7S,9R)-7-[[3-(Trifluoroacetamidomethyl)-2,3,4,6-tetradeoxy- α -L-lyxo-hexopyranosylloxyl-7,8,9,10tetrahydro-4,6,9,10,11-pentahydroxy-9-ethyl-5,12-naphtacenedione 18

mp 95-97°C; $\alpha \ln^{20}$ 192.5 (c 0.2, CHCl₃); ¹H NMR (CDCl₃); δ 1.11 (t, J=7Hz, 3H), 1.26 (d, J=6Hz, 3H), 3.19 (m, 2H), 4.16 (m, 2H), 4.87 (s, 1H), 5.14 (s, 1H), 5.47 (s, 1H), 6.53 (s, WH=11Hz, 1H), 7.30 (d, J = 6Hz, 1H), 7.68 (t, J = 8Hz, 1H), 7.78 (d, J = 6Hz, 1H), 12 (s, 1H), 12.73 (s, 1H), 13.5 (s, 1H) ppm; I.R. (CH₂Cl₂): 3680, 3600, 3520 and 3440 (OH), 1725 (C=O), 1620, 1580 and 1550 (C=C), 1290, 1265, 1230, 1200 and 1165 (C-O) cm⁻¹; HRMS: calcd. for C₂₀H₃₀O₁₀NF₃ [M+H]⁺: 610.18998; found 610.19000.

(7S,9R,10R)-7-[[3-(Trifluoroacetamidomethyl)-2,3,4,6-tetradeoxy-a-L-lyxo-hexopyranosylloxyl-7.8.9.10-tetrahydro-1.4.6.9.11-pentahydroxy-10-carbomethoxy-9-ethyl-5.12-naphtacenedione 19 mp 186°C; [α]_D²⁰-140 (c 0.11, CHCl₃); ¹H NMR (CDCl₃); δ 1.12 (t, 3H), 1.25 (d, H), 3.21 (m, 2H), 3.7 (s, 3H), 4.11 (m, 1H), 4.57 (s, 1H), 5.25 (s, 1H), 5.49 (s, 1H), 6.31 (s, 1H), 12.29 (s, 1H), 12.95 (s, 1H) ppm; HRMS: calcd. for $C_{31}H_{32}O_{12}NF_3$ 667.18763; found 667.18766.

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